

 **InterPore2026**

18th Annual Meeting &
Conference Courses

19 - 22 May 2026, Nantes, *France*
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Multi-phase Flow and Bubble Management in Anion Exchange Membrane Water Electrolysis for Green Hydrogen Production

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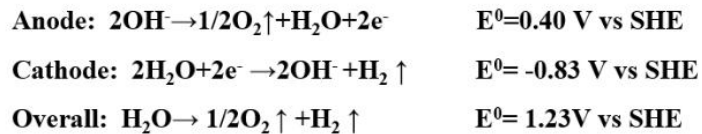
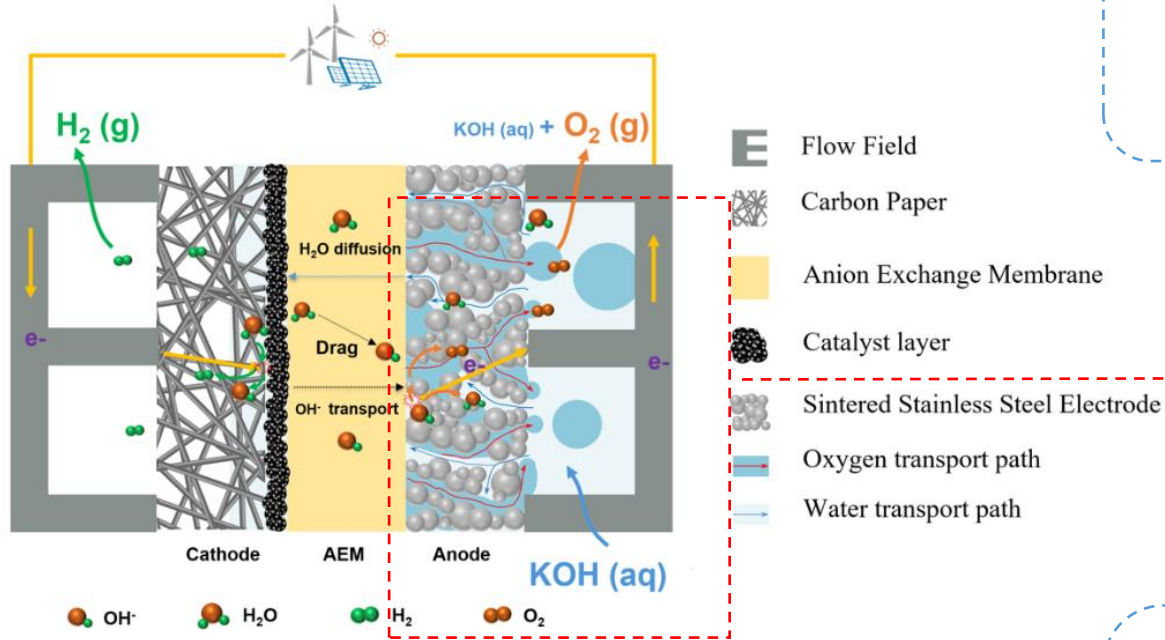
Outline

- ❑ **Background and bubble issues**
- ❑ **Methodology**
- ❑ **Pore size and thickness optimization**
- ❑ **Hierarchical bubble-management design**
- ❑ **Summary**

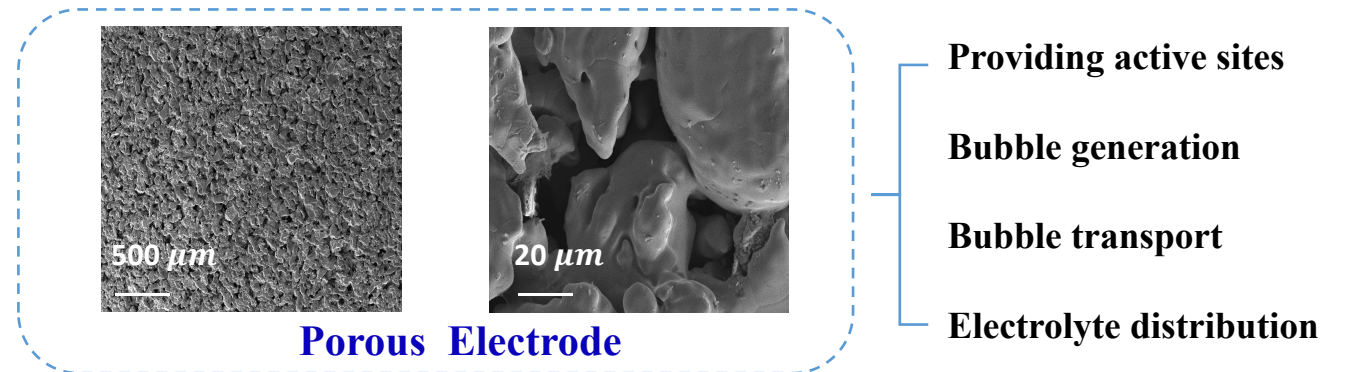
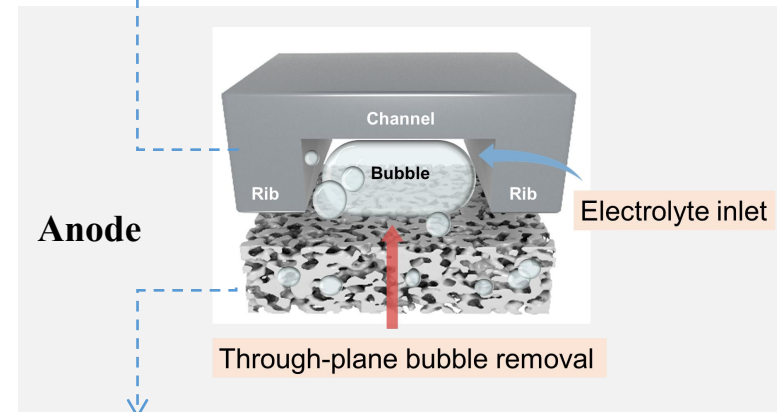
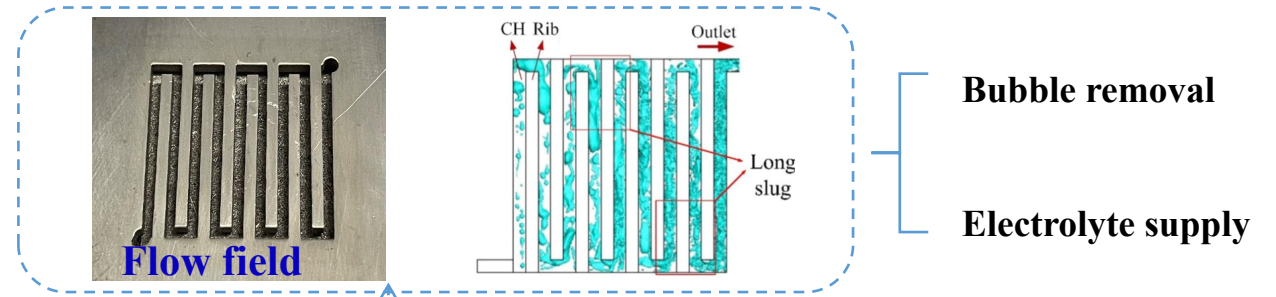
Background

➤ Water electrolysis for green hydrogen production

Taking the AEMWE as an example:

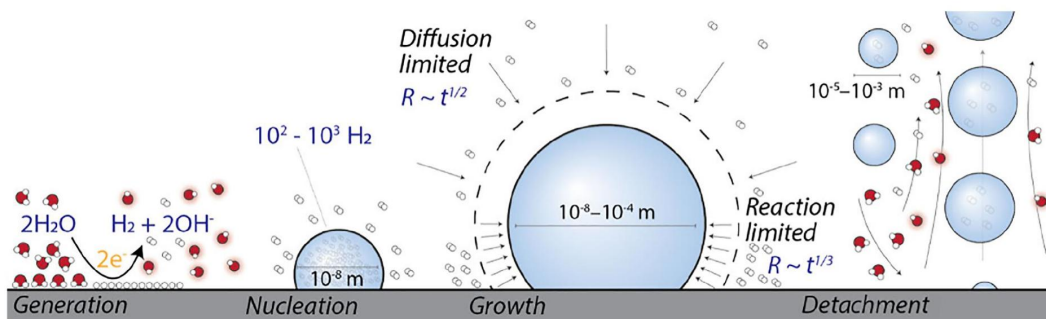
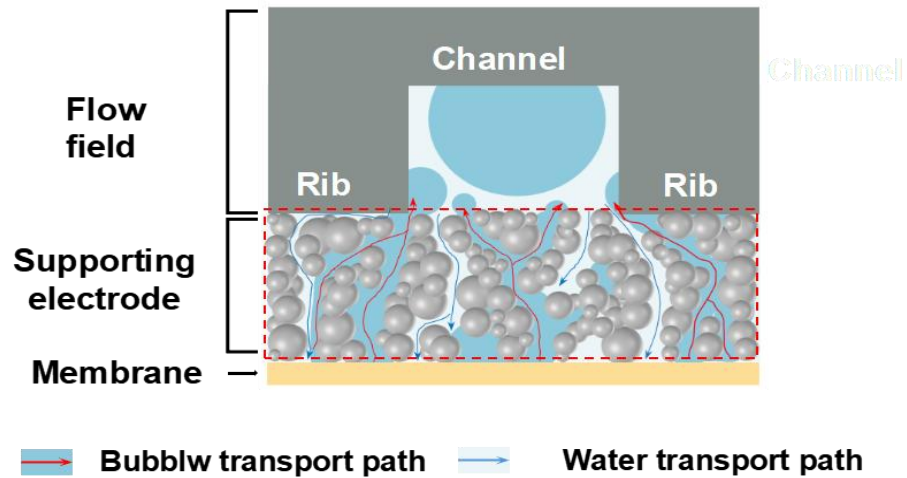


Anion Exchange Membrane Water Electrolysis



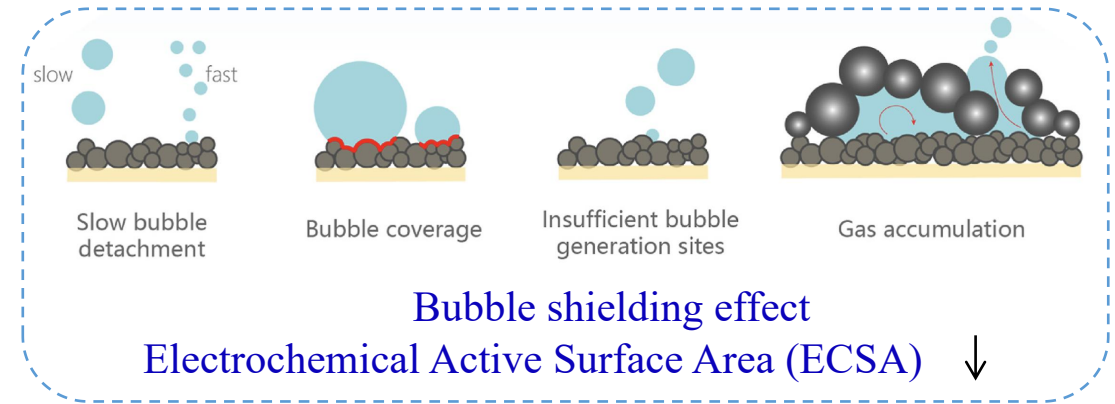
Bubble issues

➤ Bubble issues in the porous electrode

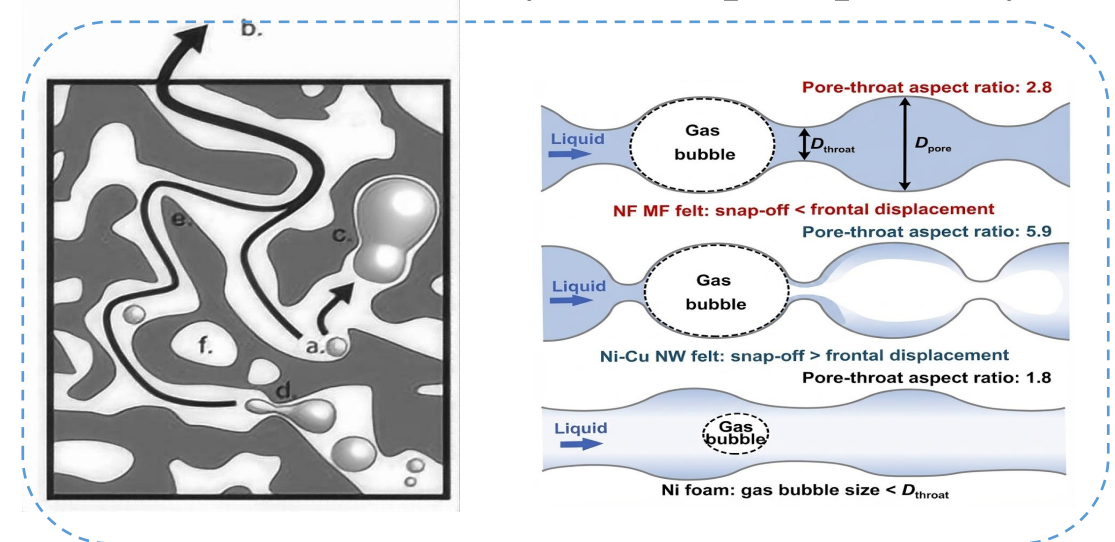


Bubble life cycle

❑ Bubbles cover the active electrode surface

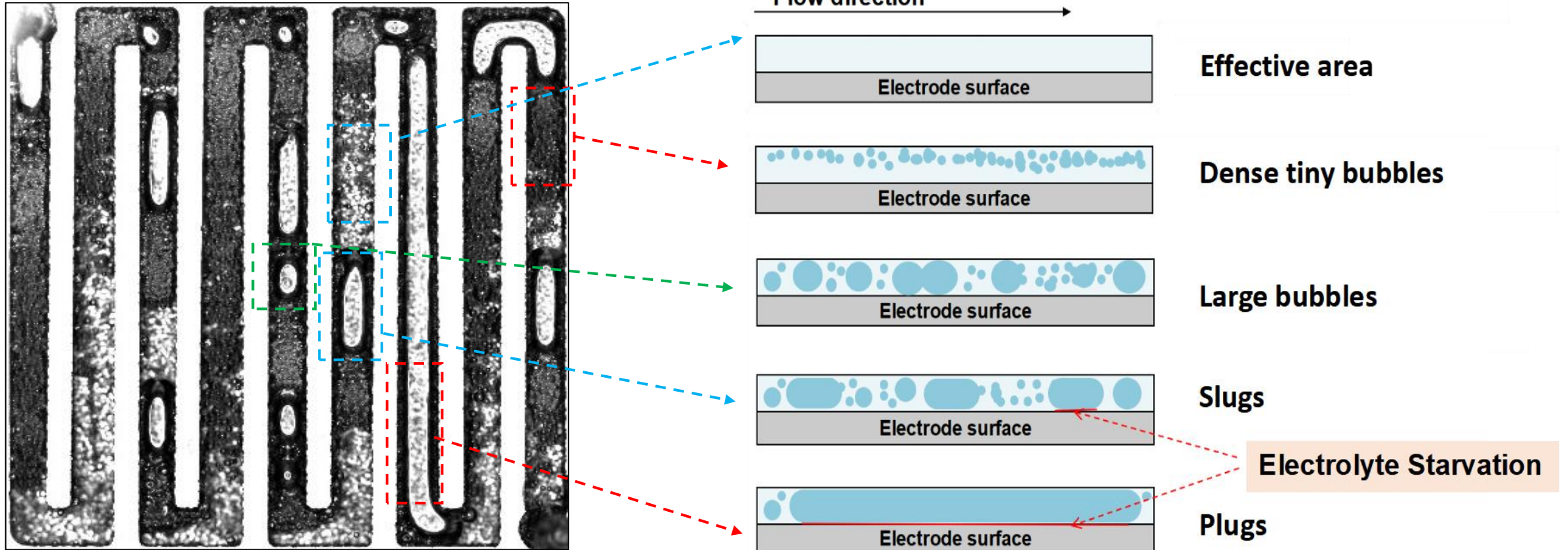


❑ Bubbles block electrolyte transport pathways



Bubble issues

➤ Bubble issues in the flow field



Two-phase flow in the flow field

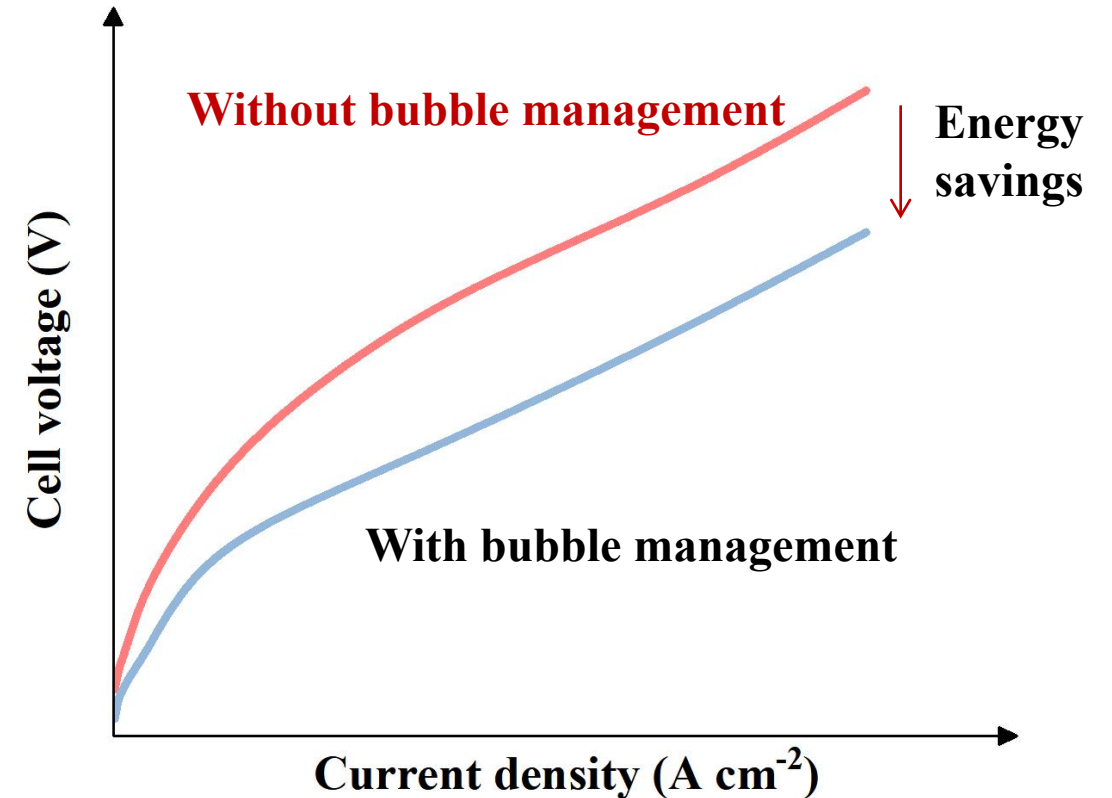
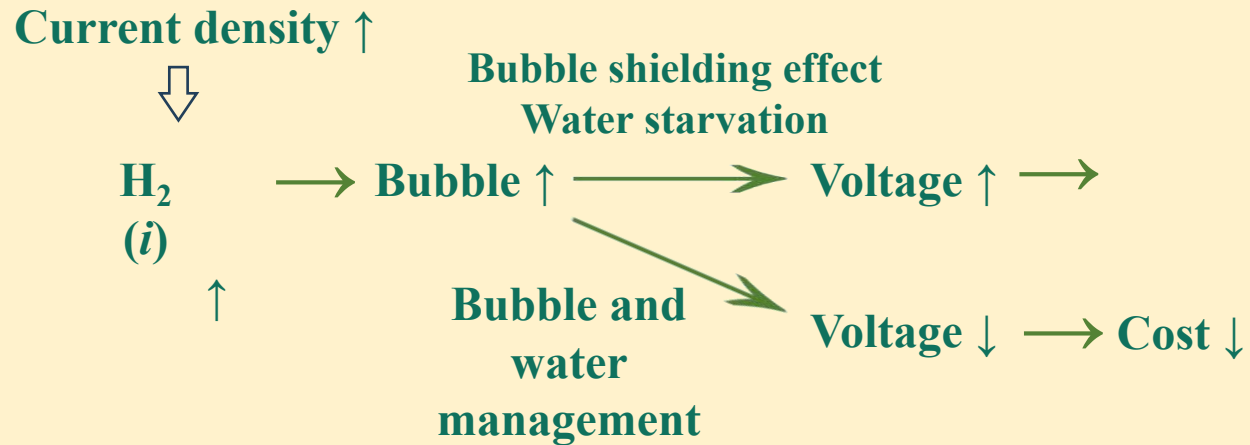
- Bubble coalescence and blockage restrict liquid transport pathways, leading to electrolyte starvation

Bubble issues

➤ Why bubble management matters?

How to maximize H_2 production and minimize cost?

Hydrogen production at high current densities:

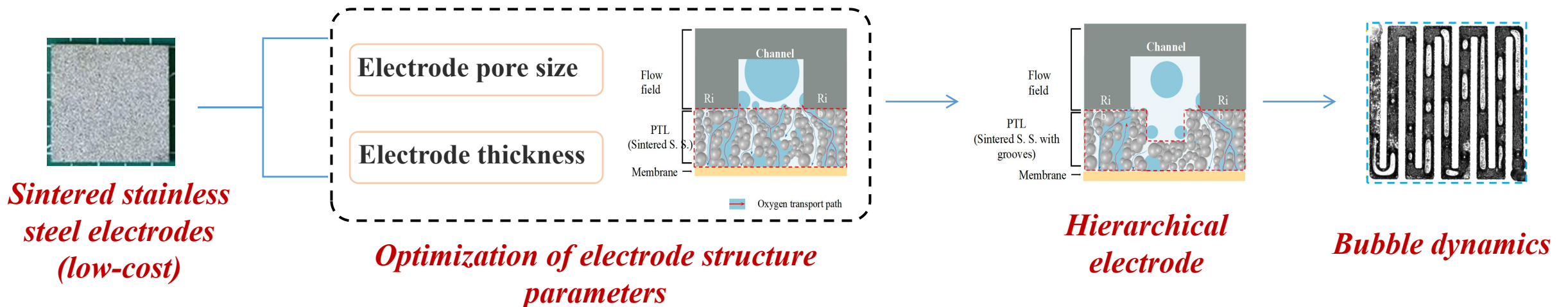


- Effective bubble management lowers the cell voltage at high current densities, enabling energy savings during hydrogen production

Bubble issues

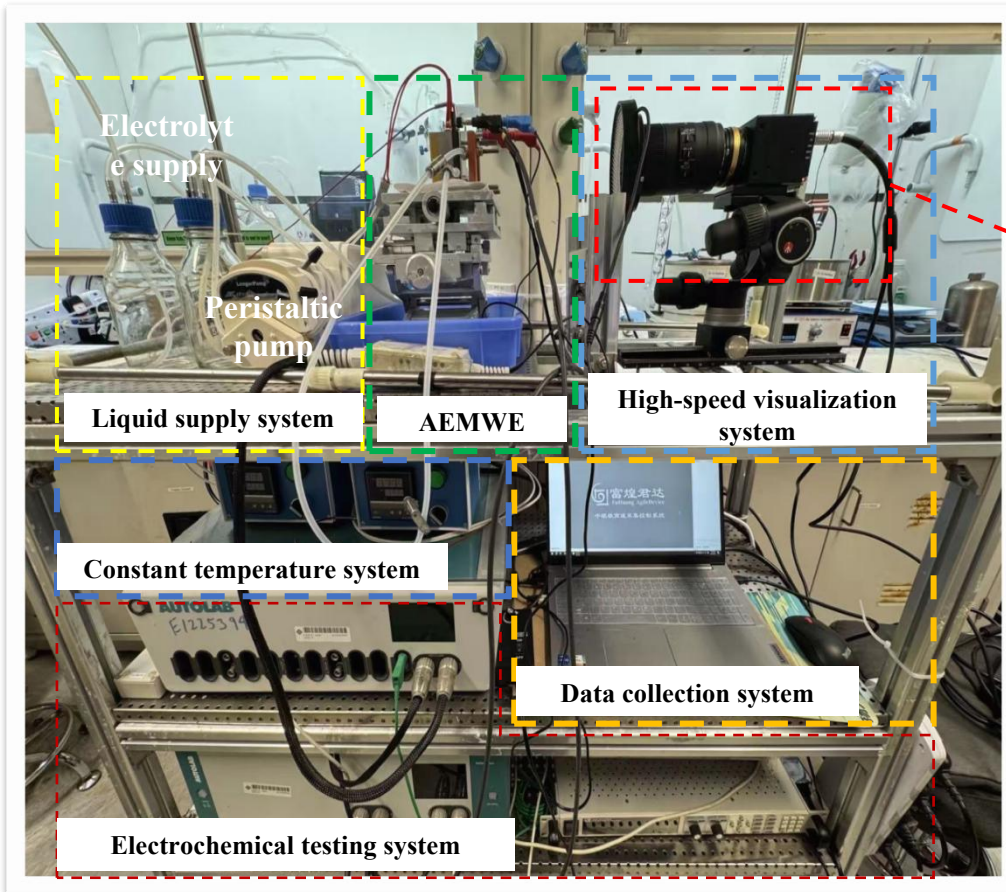
➤ Research objective

- ❑ To investigate the impact of electrode structural parameters on electrochemical performance and mass transport via numerical and experimental methods
- ❑ To elucidate bubble dynamics and mass-transfer mechanisms at high rates via in-situ visualization
- ❑ To design hierarchical grooves for improved bubble evacuation and liquid supply



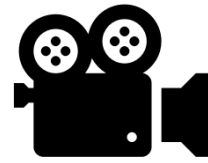
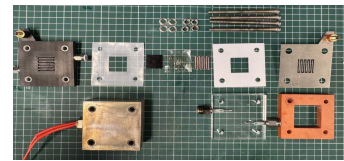
Methodology

➤ Electrochemical testing platform and in-situ visualization system

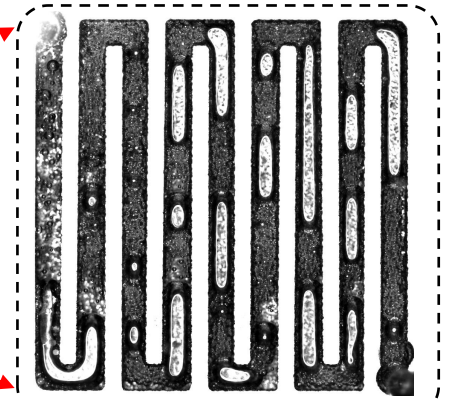
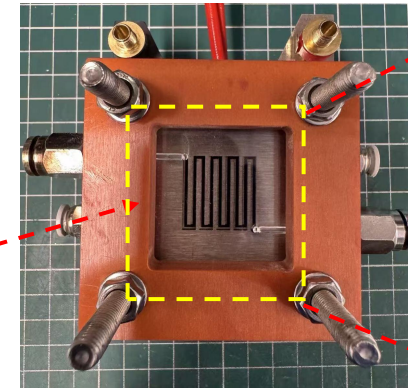


Electrochemical testing platform

Visualize bubble behavior in flow fields



High-speed camera

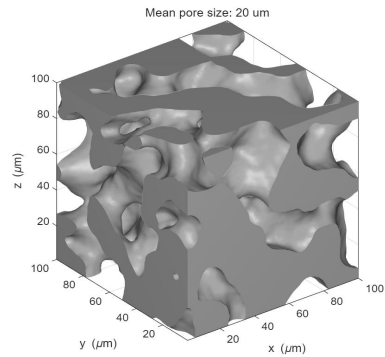


Operating condition and components

- **Operating Temperature:** 80 °C; Electrolyte: 1 M KOH
- **Electrolyte flow rate:** 5.0 mL min⁻¹;
- **Anode CL and PTL:** Sintered S. S. electrode;
- **Cathode CL and GDL :** Carbon paper coated with 0.5 mg cm⁻² Pt/C;
- **Membrane:** Nafion 115
- **Flow field plate:** Graphite plate(cathode), Nickel plate(anode)

Methodology

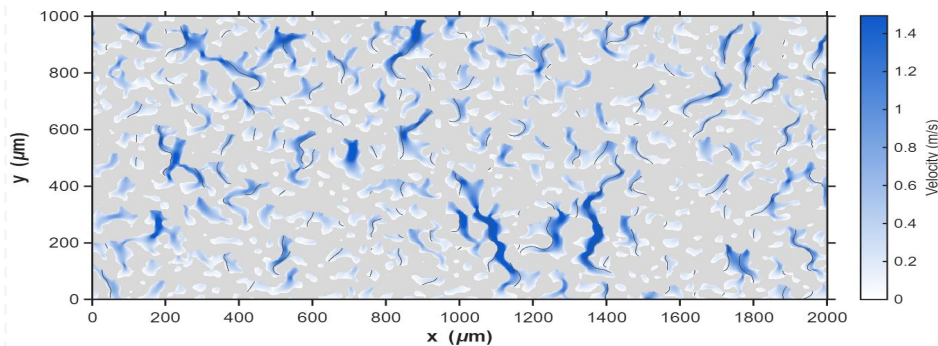
➤ Pore-scale numerical simulation of porous electrode



Size:
 $100 \times 100 \times 100 \mu\text{m}^3$
Lattice:
 $80 \times 80 \times 80$ nodes
($1.25 \mu\text{m}/\text{grid}$)

REV (Gaussian Random Field)

Study on the effect of electrode pore size



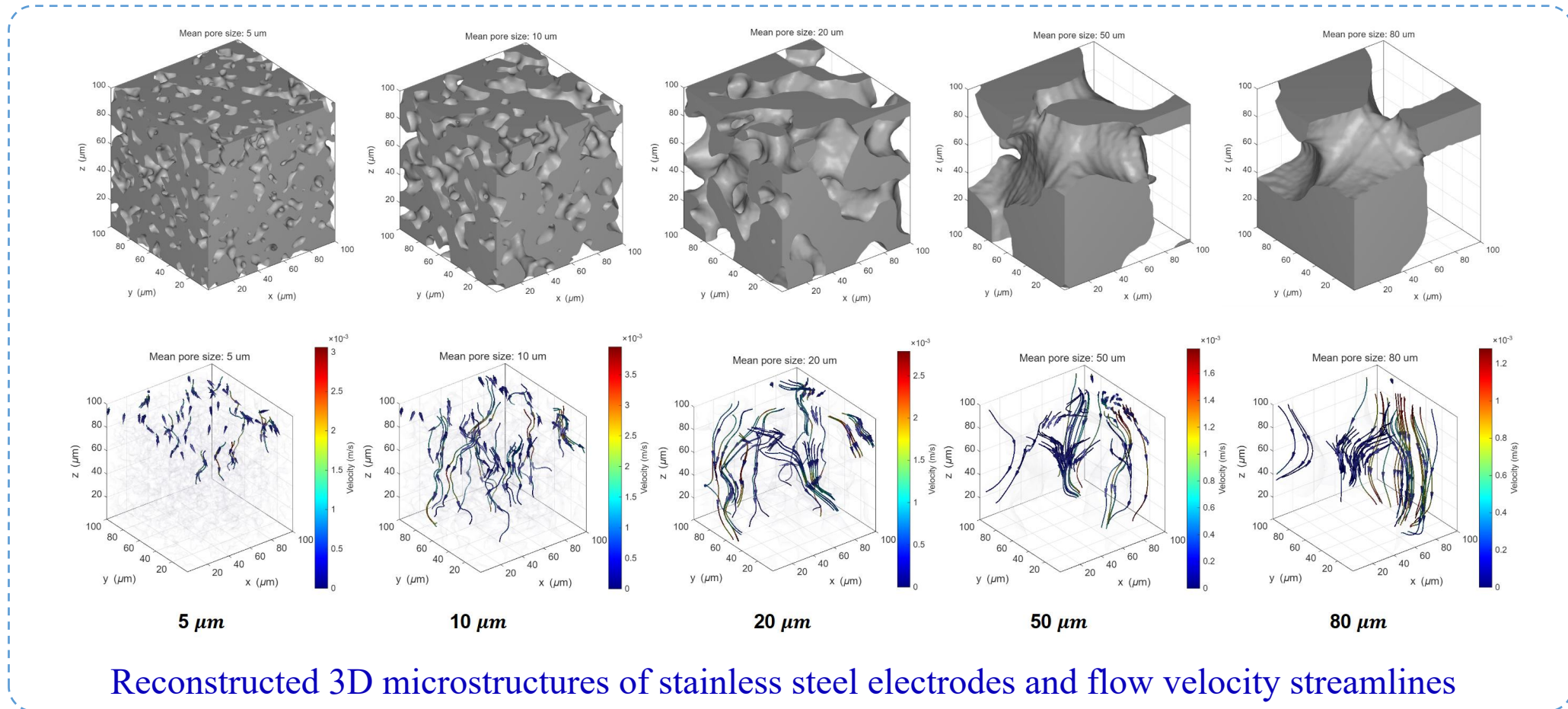
Study on the effect of electrode thickness

Transport properties and electrochemical properties

- ✓ Permeability
- ✓ Pressure drop (ΔP)
- ✓ Tortuosity
- ✓ Specific surface area (SSA)
- ✓ Area specific resistance (ASR)
- ✓ The local effective active area density
- ✓ Effective reaction area utilization (U)

Research results

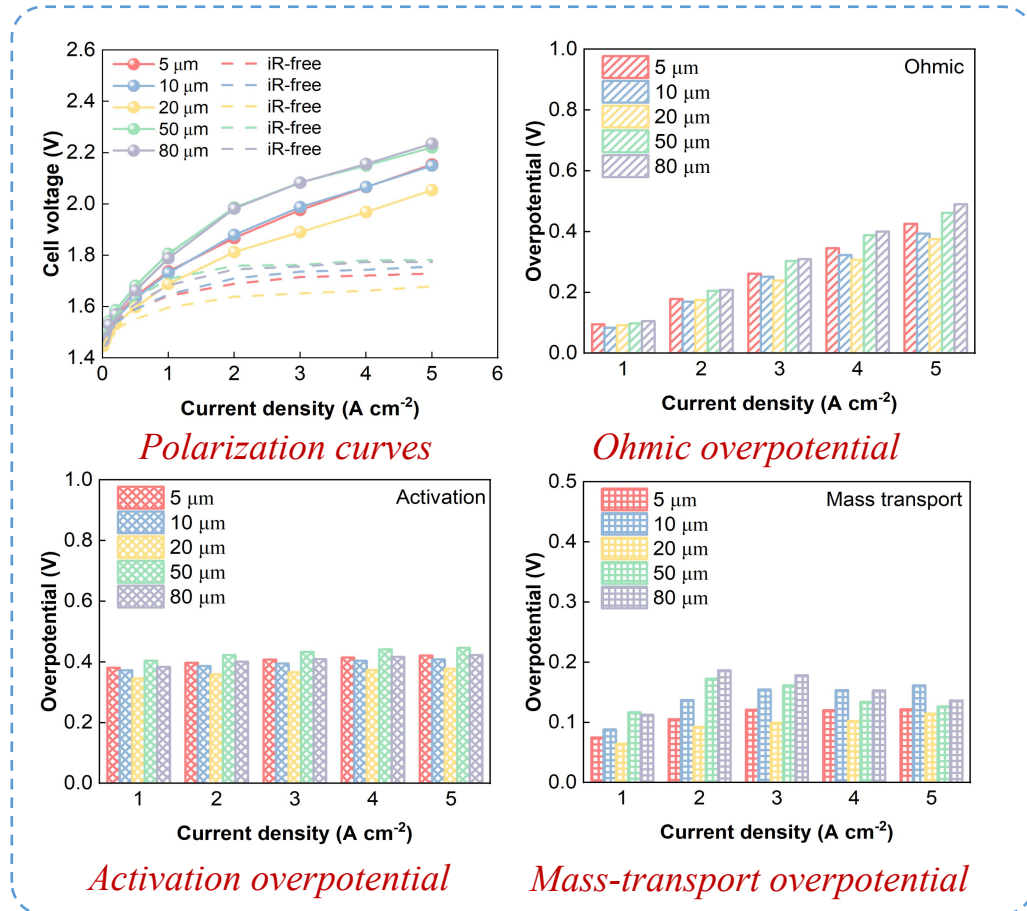
➤ Pore-size-dependent flow transport in porous electrodes



- Pore-scale simulations identify 20 μm as the optimal pore size, balancing pore connectivity, flow uniformity, and transport resistance.

Research results

➤ Pore-size-dependent cell performance and loss breakdown



Key finding:

- 20 μm shows the lowest cell voltage by balancing active surface area and transport resistance

Small pores → poor liquid/bubble transport

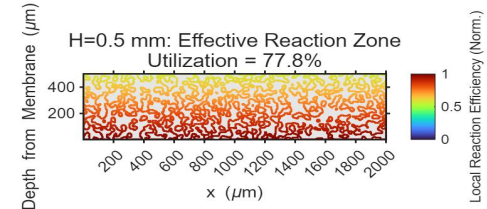
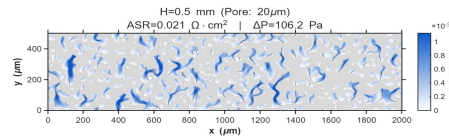
Large pores → reduced surface area/contact quality

- An intermediate pore size minimizes cell voltage by balancing active surface area, contact resistance, and mass-transport loss

Research results

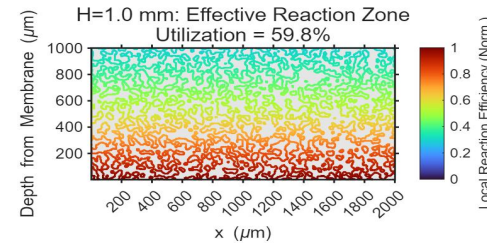
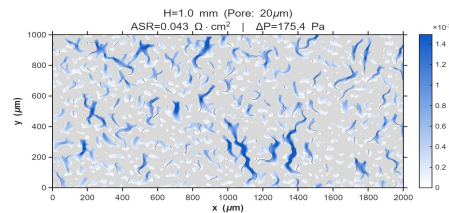
➤ Thickness-dependent flow transport in porous electrodes

0.5 mm



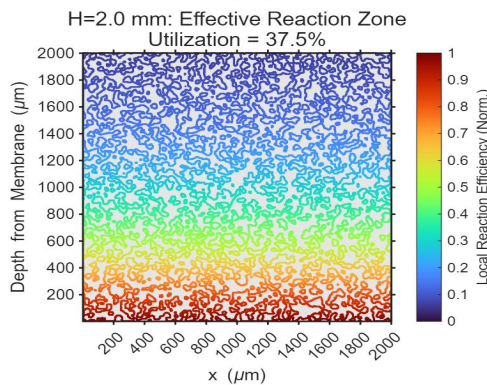
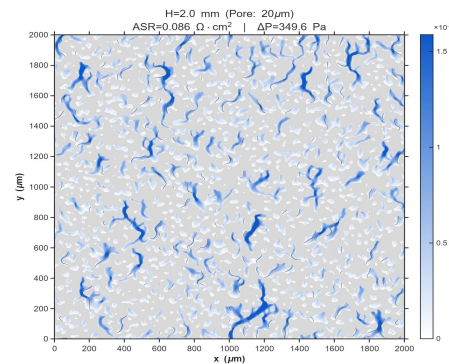
77.8%

1.0 mm



59.8%

2.0 mm



37.5%

Electrolyte flow velocity distribution

Distribution of the effective reaction zone

Thickness \uparrow

\rightarrow longer pathway

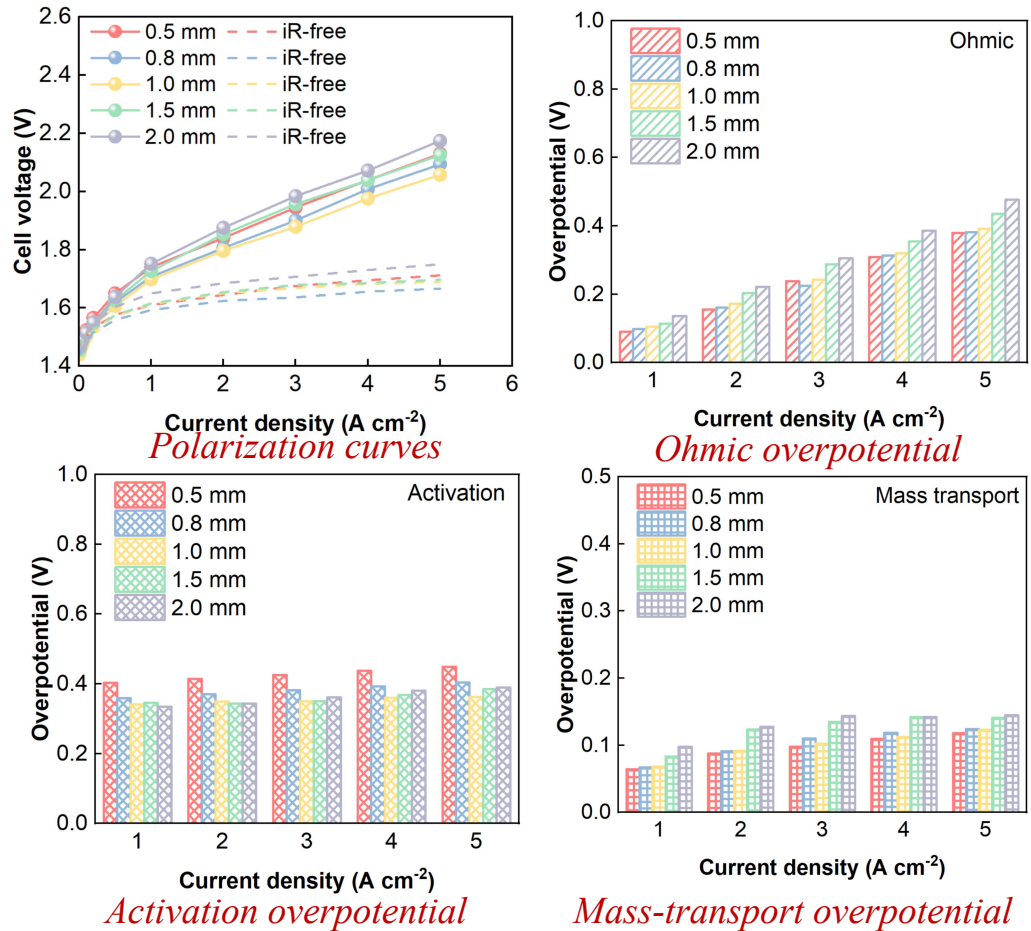
\rightarrow higher pressure drop

\rightarrow lower ECSA utilization

- Excessive electrode thickness extends transport pathways, promotes gas accumulation, and reduces effective reaction-zone utilization

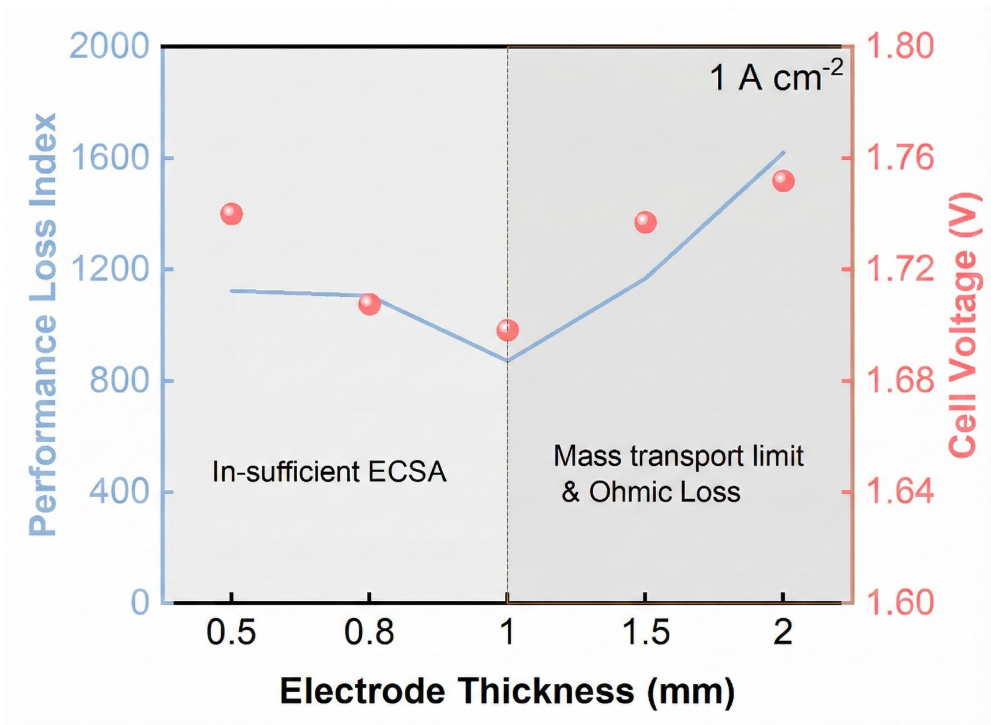
Research results

Thickness-dependent cell performance and loss breakdown



> Performance Loss Index (PLI):

$$PLI = \frac{1}{(ECSA \cdot U)^2} + 40\sqrt{j(i)} \cdot (ASR \cdot \Delta P)$$

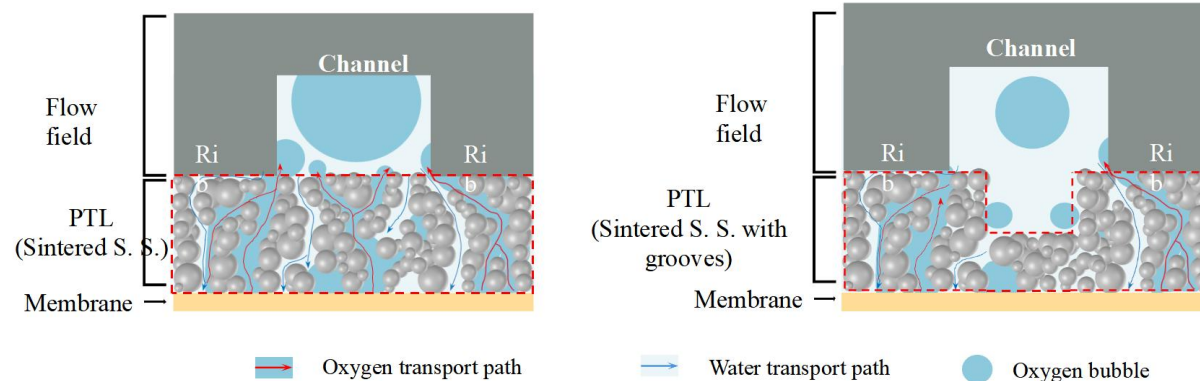
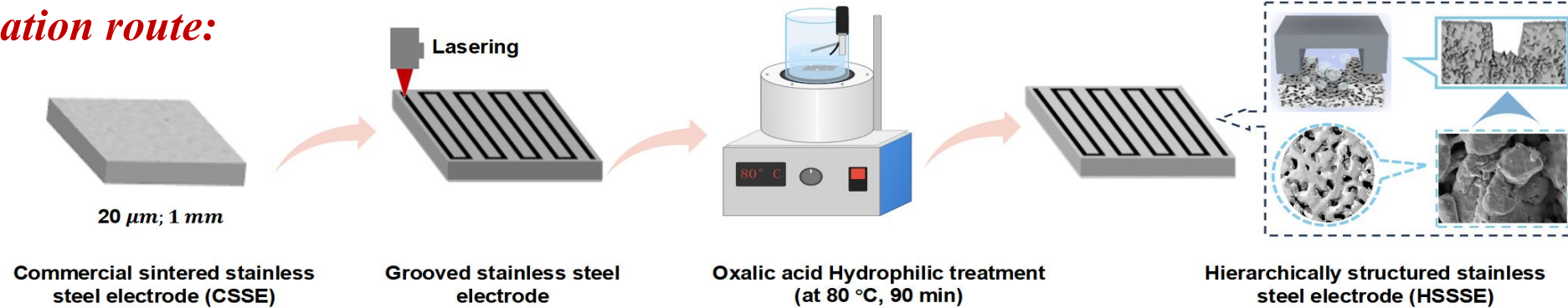


- The 1.0 mm electrode achieves the lowest cell voltage by balancing active surface area, ohmic resistance, and mass-transport loss.

Research results

Hierarchical grooves for guided bubble removal

Fabrication route:

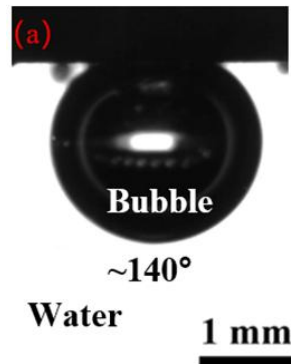


Decoupling bubble removal from liquid supply

- Hierarchical grooves are introduced into the optimized $20\ \mu\text{m}$, $1.0\ \text{mm}$ electrode to create preferential bubble-removal pathways while maintaining liquid supply.

➤ Hierarchical grooves accelerate bubble evacuation in porous electrodes

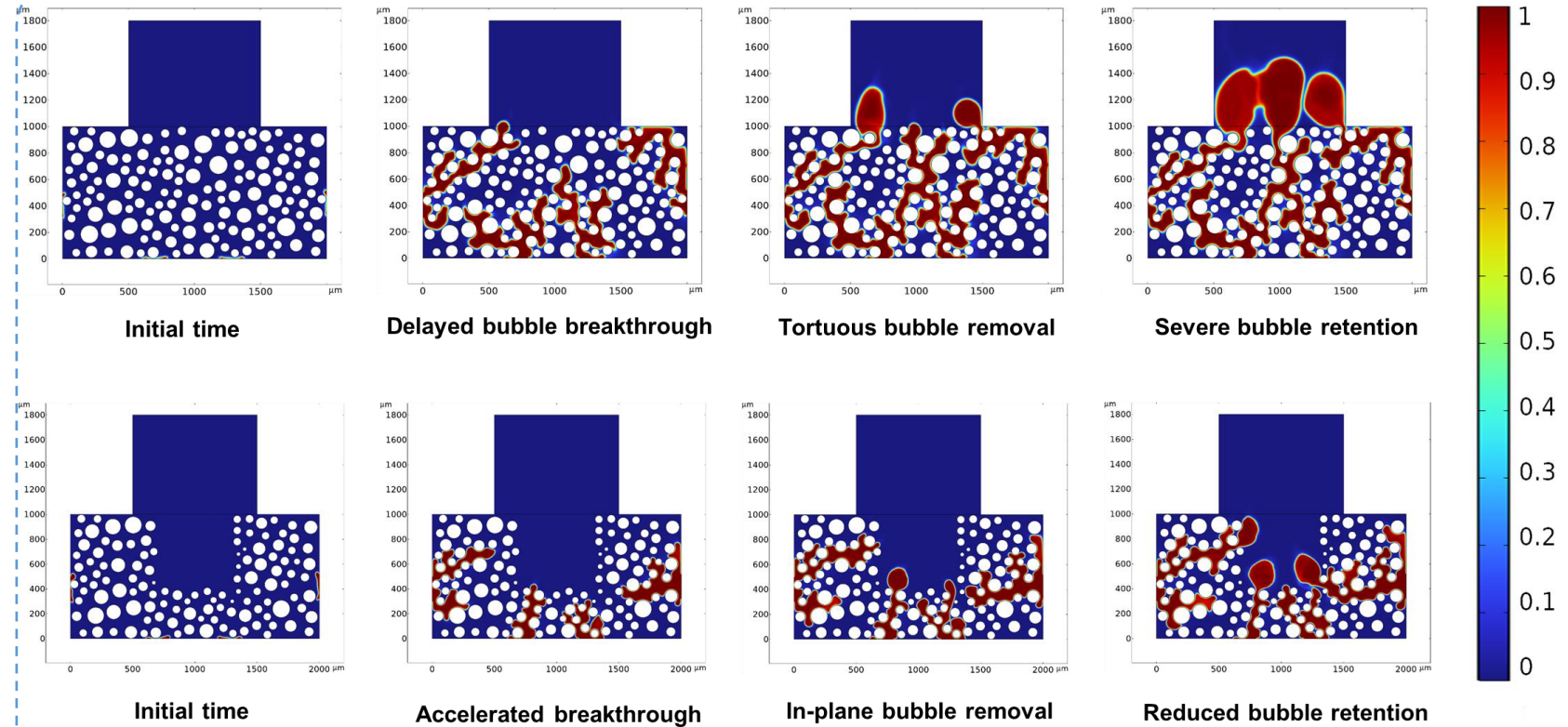
Conventional electrode



Hierarchical electrode



Underwater bubble contact angle indicates weaker bubble adhesion

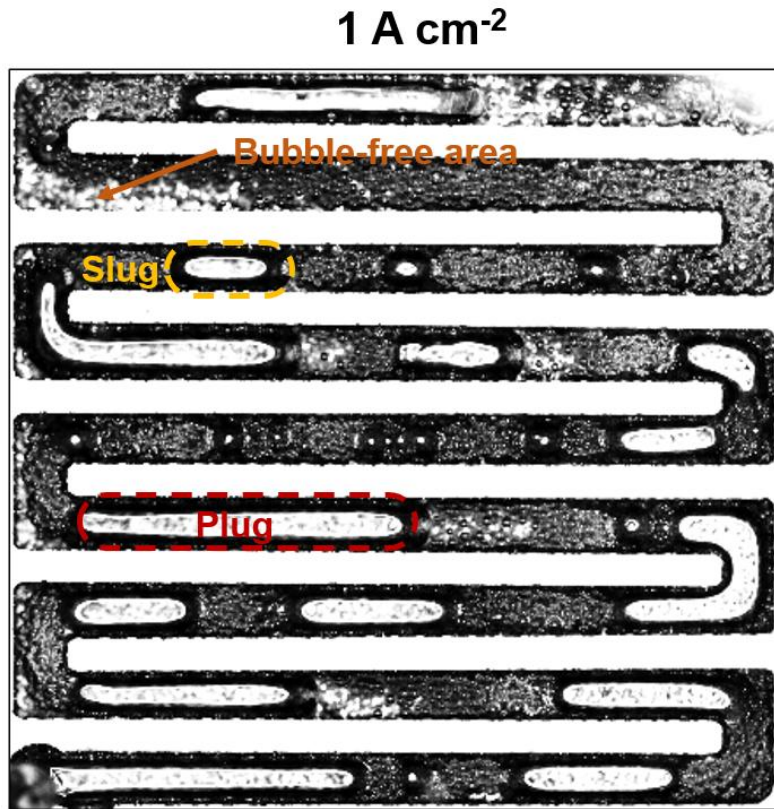


Bubble evacuation in porous media

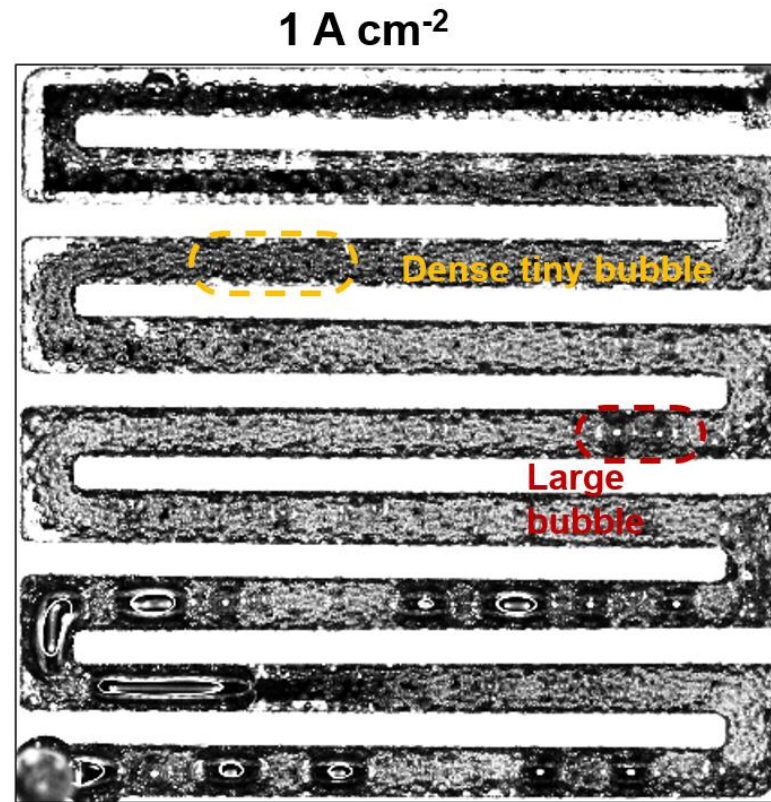
- Hierarchical grooves weaken bubble adhesion and accelerate bubble evacuation, reducing bubble retention inside the porous electrode.

Research results

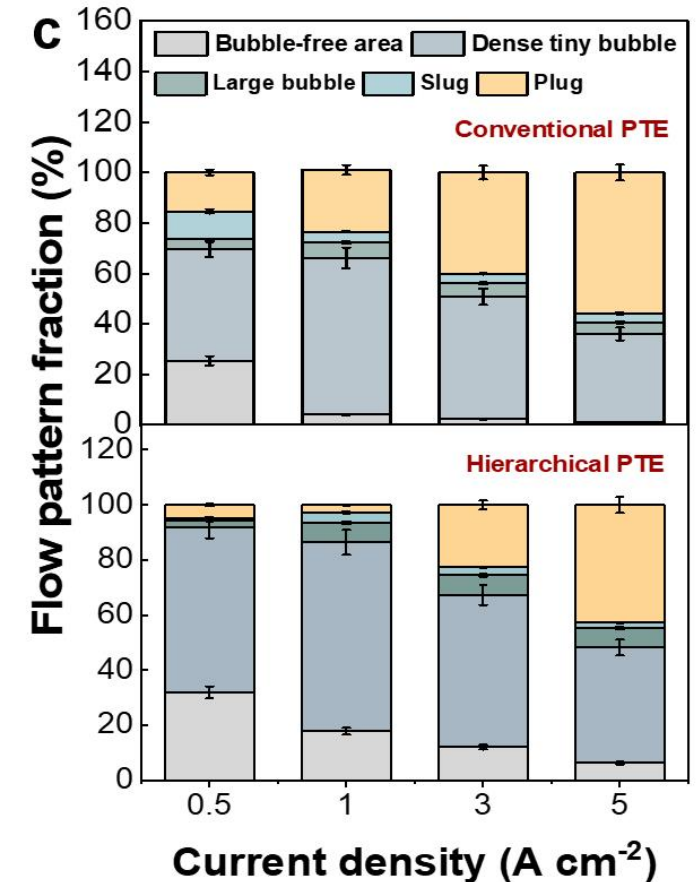
➤ Flow-field visualization confirms enhanced bubble removal



(a) Conventional PTE



(b) Hierarchical PTE



- The hierarchical electrode suppresses slug and plug flow, thereby maintaining liquid-accessible pathways in the flow field

Summary

- **20 μm pore size:** balances pore connectivity, active surface area, and transport resistance
- **1.0 mm electrode thickness:** achieves the lowest cell voltage by balancing active area and transport losses
- **Hierarchical grooves:** accelerate bubble evacuation and reduce bubble retention inside the porous electrode
- **Flow-field visualization :** confirms suppressed slug/plug flow and improved liquid accessibility

Acknowledgements

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Thank you!



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