

Pore-scale Simulations On The Impacts Of Hydrate Production Approaches On Gas And Water Transport In Hydrate-bearing Sediments

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Outline

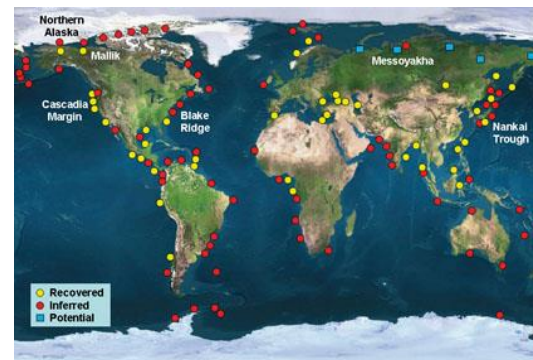
- Background and motivation
- Lattice Boltzmann(LB) method
- LB model for multiphase reactive transport processes
 - Model verification
 - Simulation study on relative permeability estimation during hydrate dissociation
- Conclusion

Background and motivation

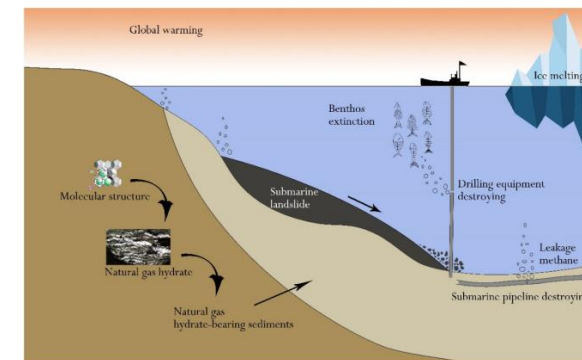
- Gas hydrates resource
 - Hydrates are crystalline compounds composed of gas molecules trapped in a lattice of water molecules
 - Stable under low temperature and high pressure
 - Over 3000 trillion cubic meters
- Production challenges associated with permeability changes
 - Gas surge in production
 - Excessive water production
 - Economic feasibility



(Wikipedia: Methane hydrate)



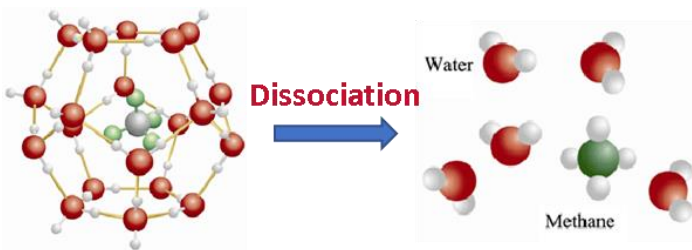
(Boswell et al. , 2009)



(Ren, X. et al., 2020)

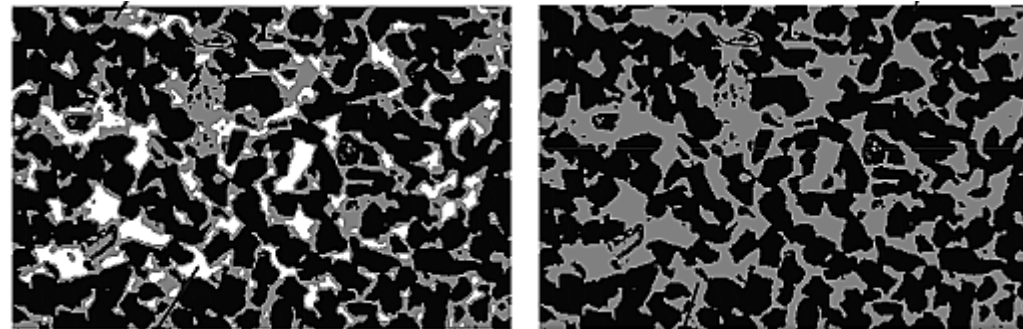
Hydrate dissociation processes

- Essential physicochemical processes during hydrate dissociation
 - Dissociation reaction of methane hydrate: $\text{CH}_4 \cdot 5.75\text{H}_2\text{O} \rightarrow \text{CH}_4 \uparrow + 5.75\text{H}_2\text{O}$
 - Triggered by changes in temperature/pressure conditions
 - Endothermic, self-inhibition reaction
 - Evolution of pore structure during dissociation
 - Fluids transport through the interconnected network of pores
 - Heat transfer from artificial heat source/ sensible heat in adjacent layers



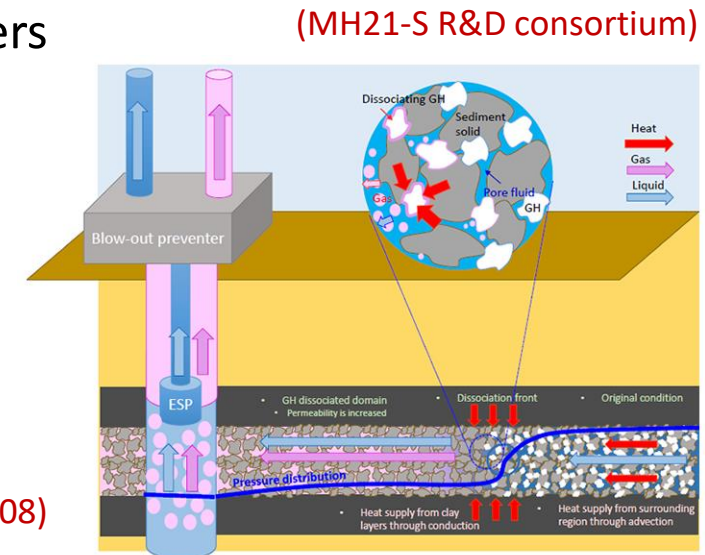
(Lee, J. Y. et al., 2011)

Dissociation



White: hydrate; **Black:** sand; **Grey:** void space (Malinverno et al., 2008)

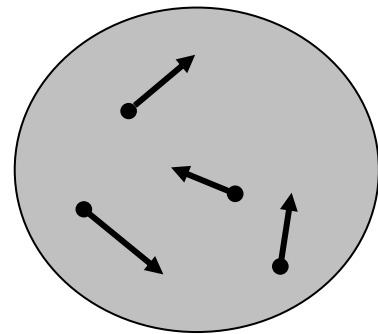
Evolution of pore structure



Hydrate production

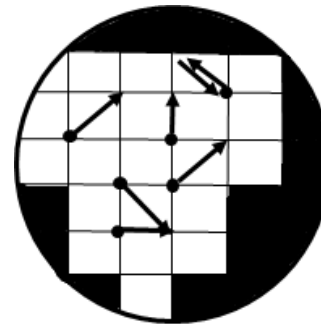
Methodology selection

- Lattice Boltzmann simulation for reactive transport and permeability estimation
 - Flexibility in boundary conditions treatment
 - Handle complex pore geometries
 - Capture pore scale interactions between fluids and solids
 - Reliable numerical stability
 - Ease of implementation and efficient parallelization



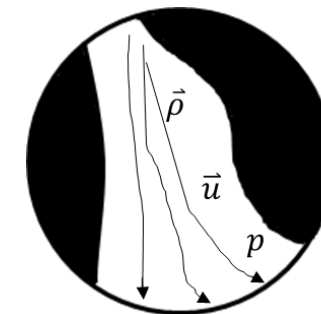
Molecular Dynamics
molecule movement

Simplified
kinetic models



LB method
particle distributions

averaged
fluid quantities



N-S equation
macroscopic quantities

Lattice Boltzmann model

- Multiphase hydrodynamic LB model

- Phase field multiphase model (Jacqmin, D, et al., 1999)

- Derived by minimizing the Gibbs free energy of the multiphase system to predict phase interface evolutions
 - Applies the advection-diffusion form of equation to describe the evolution of phase index
 - Introduces diffuse interfaces between different phases to achieve a smooth transition of physical properties

- Phase field LB model (Geier, M., et al. 2015)

Phase index distribution

Pressure distribution

$$\begin{aligned}
 h_i(\mathbf{x} + \mathbf{e}_i \Delta t, t + \Delta t) &= h_i(\mathbf{x}, t) - \frac{h_i(\mathbf{x}, t) - \overline{h}_i^{eq}(\mathbf{x}, t)}{\tau_\varphi} + S_i^\varphi(\mathbf{x}, t) \Delta t \\
 f_i(\mathbf{x} + \mathbf{e}_i \Delta t, t + \Delta t) &= f_i(\mathbf{x}, t) - \frac{f_i(\mathbf{x}, t) - \overline{f}_i^{eq}(\mathbf{x}, t)}{\tau_f} + S_i(\mathbf{x}, t) \Delta t
 \end{aligned}$$

$\rho(\varphi)$
 $F_{tot,i}(\varphi)$

- Source term due to dissociation and forces (Verdier, W., et al., 2020)

$$S_i^\varphi(\mathbf{x}, t) = F_{tot,i}^\varphi + \dot{m}_i^\varphi$$

Ensure the surface tension is correctly accounted at interfaces

Mass source contribution to phase index

$$S_i(\mathbf{x}, t) = F_{tot,i} + \dot{m}_i$$

Body force, viscous force, surface tension

Mass source contribution to pressure

Lattice Boltzmann model

- Multiphase hydrodynamic LB model
 - Equivalent macroscopic multiphase model (Geier, M., et al. 2015)

$$\frac{\partial \rho}{\partial t} + \nabla \cdot (\rho \mathbf{u}) = \dot{m}$$

$$\frac{\partial \rho \mathbf{u}}{\partial t} + \nabla \cdot (\rho \mathbf{u} \mathbf{u}) = -\nabla p + \nabla \cdot (\rho \vartheta (\nabla \mathbf{u} + (\nabla \mathbf{u})^T)) + F_{\text{tot}} + \dot{m} \mathbf{u}$$

} Hydrodynamic equations

$$\frac{\partial \varphi}{\partial t} + \nabla \cdot (\varphi \mathbf{u}) = \nabla \cdot \left(M_\varphi \left(\nabla \varphi - \frac{4}{W} \varphi (1 - \varphi) \mathbf{n} \right) \right) + \frac{\dot{m} \varphi}{\rho}$$

} Volume diffusive term

Phase index equation
(Allen-Cahn equation)
(Allen, S. M., et al., 1979)

Lattice Boltzmann model

- Heat transfer LB model

- Heat transfer equation

Temperature distribution

$$g_i(\mathbf{x} + c\mathbf{e}_i\Delta t, t + \Delta t) = g_i(\mathbf{x}, t) - \frac{1}{\tau_{g,k}} [g_i(\mathbf{x}, t) - g_i^{eq}(\mathbf{x}, t)] + \Delta t F_i$$

- Heat source due to latent heat of dissociation and conjugate condition treatment (Karani et al., 2015) (Moridis, 2012)

$$F_i = \omega_i (S_{conj} + S_{latent})$$

- Equivalent macroscopic heat transfer equation (Karani et al., 2015)

$$\rho h \frac{\partial T}{\partial t} + \nabla \cdot (\rho h \mathbf{u} T) = \nabla \cdot (k \nabla T) + S_{latent}$$

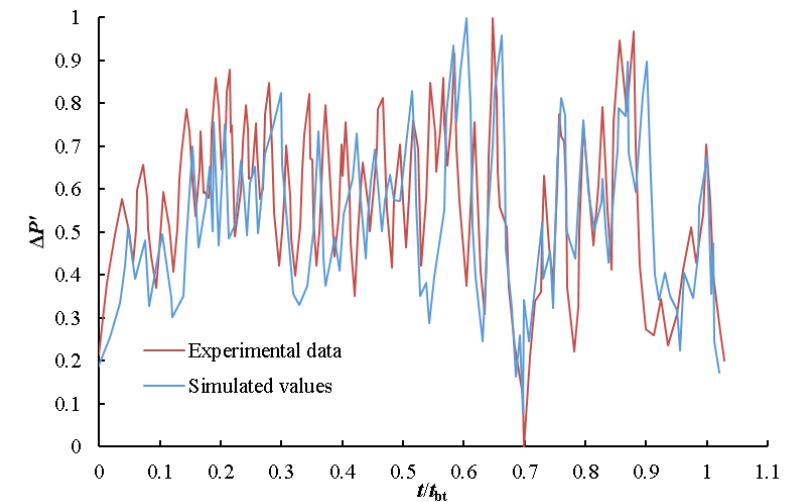
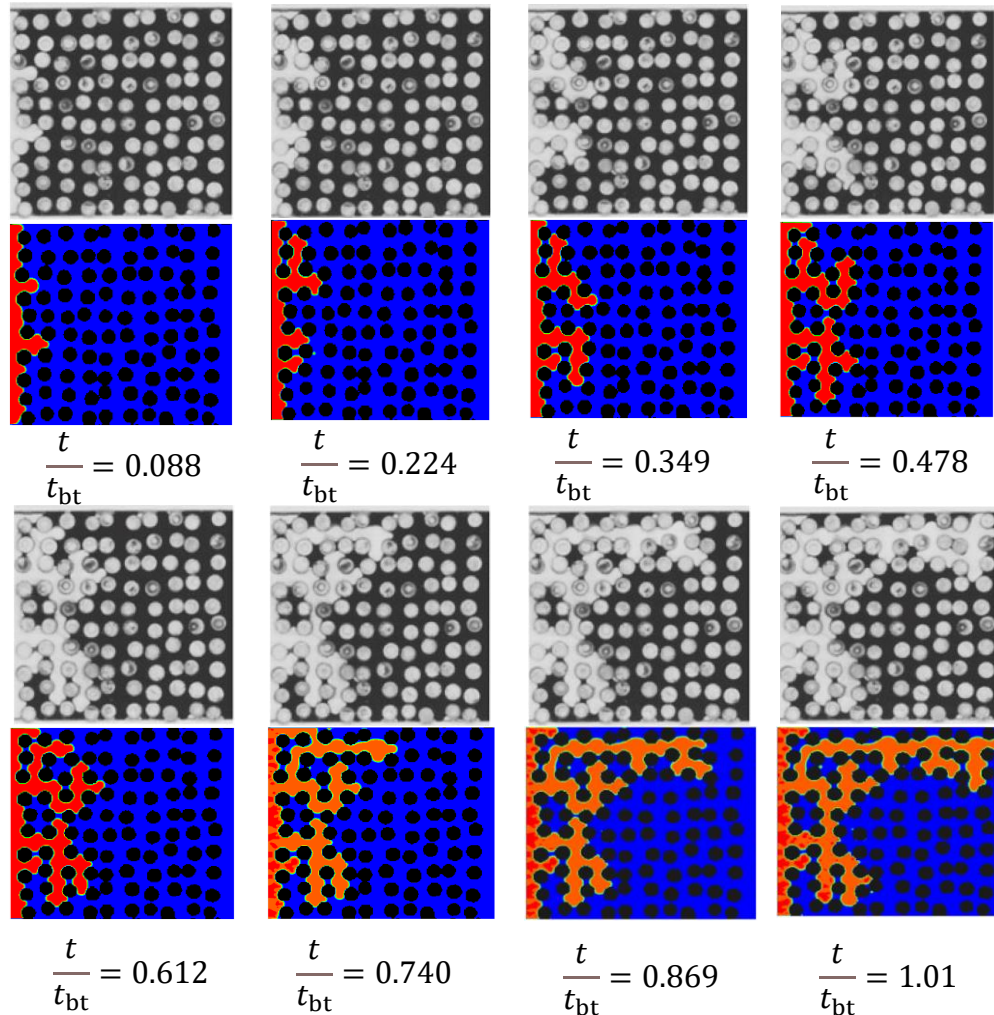
- Geometry alteration model (volume of pixel) (Kang et al., 2006)

VOP

$$V_s(t + \Delta t) = V_s(t) - AV_m S_{reaction}$$

Multiphase transport model verification

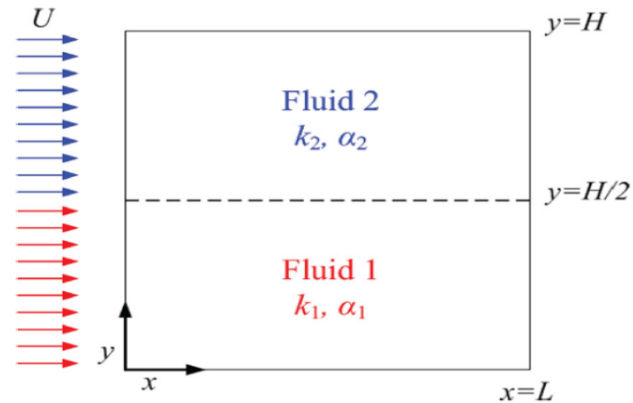
- Verification on hydrodynamic processes
 - Multiphase flow in porous medium (Aursjø et al., 2011)



Pressure comparisons

Heat transfer model verification

- Verification on heat transfer processes

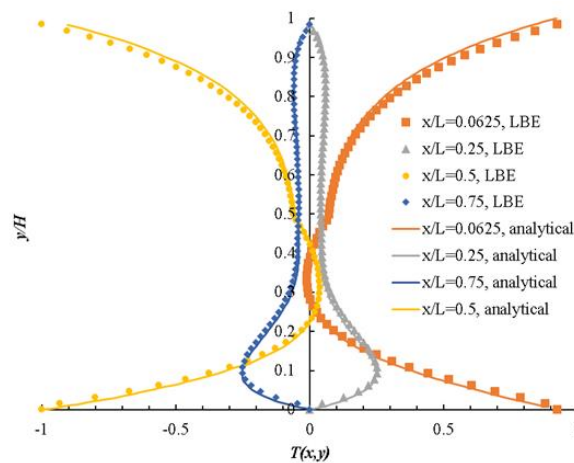


Horizontal walls have fixed sinusoidal temperatures:

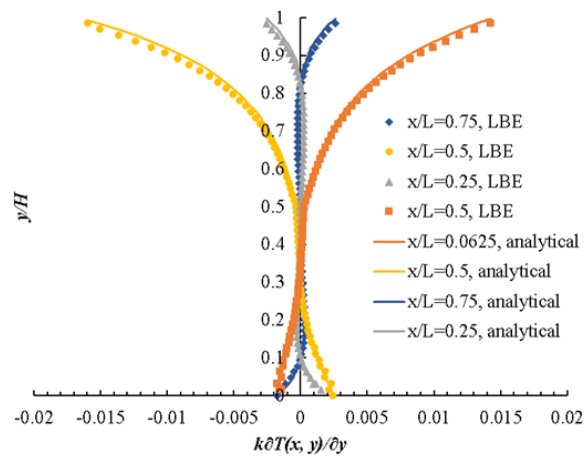
$$T(x, y = 0) = T(x, y = H) = \cos(\omega x), \omega = 2\pi/L$$

(Karani et al., 2015)

Steady-state convection-diffusion heat transfer



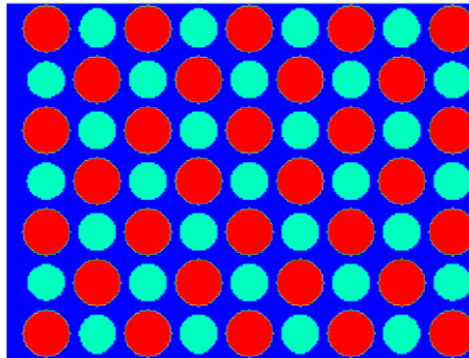
y/H (vs) T



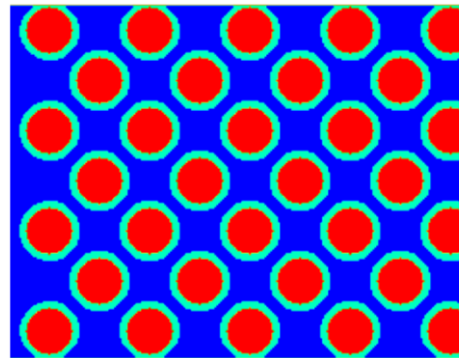
y/H (vs) $k\delta T(x, y)/\delta y$

Simulation for gas and water transport processes

- Simulation setup
 - Evolution of pore structure defines the relative permeability(K_r) vs wetting-phase (water) saturation (S_w) relation
 - Coupling between fluid transport and heat transfer plays a dominant role in pore structure evolution
 - Study the coupling effects on the $K_r - S_w$ relations on homogeneous hydrate-bearing sediment with two common distribution morphologies under different dissociation conditions



Pore-filling

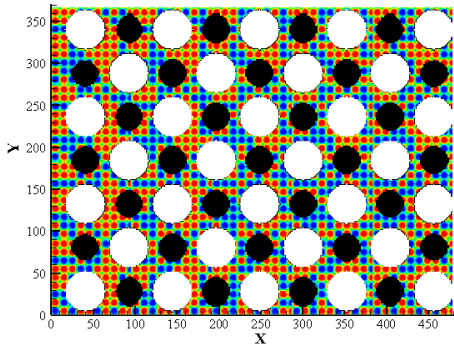


Grain-coating

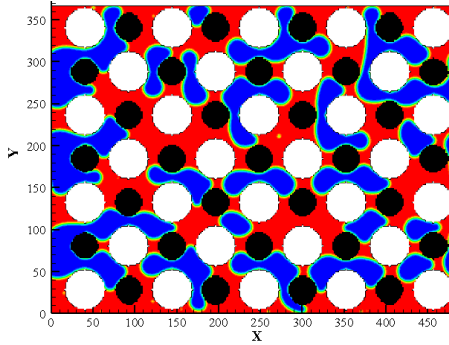
$S_{\text{hyd}}=35\%$
2.42 mm \times 1.84 mm
484 \times 368
P=12MPa
T=287.75 K

Simulation for gas and water transport processes

- Relative permeability estimation during hydrate dissociation
 - Baseline case: isothermal LB simulation on fluid transport
 - Pore structure evolution during dissociation is modeled by assumed evolution patterns
 - Boundary conditions
 - Periodic boundary condition on outer boundaries
 - Bounce-back boundary condition on fluid-solid interfaces
 - Wettability boundary condition on solid boundaries (water contact angle on solid interface: 0°) (Dai, S., et al., 2019)

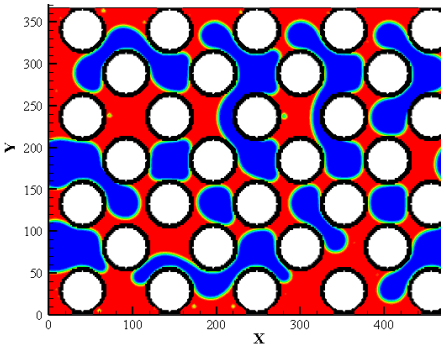


Initialization

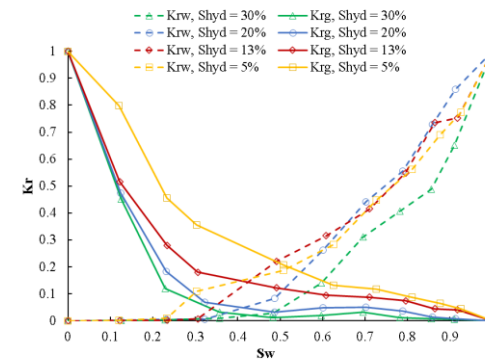


$S_{hyd}=20\%$

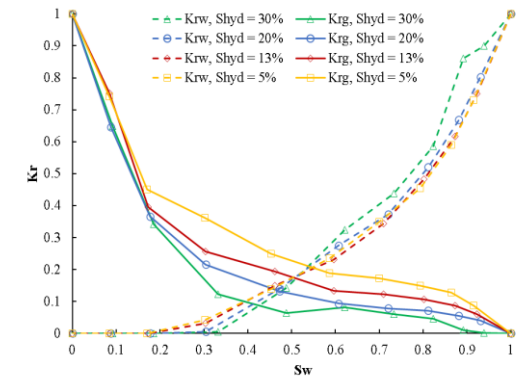
Pore filling



Grain-coating



Pore filling

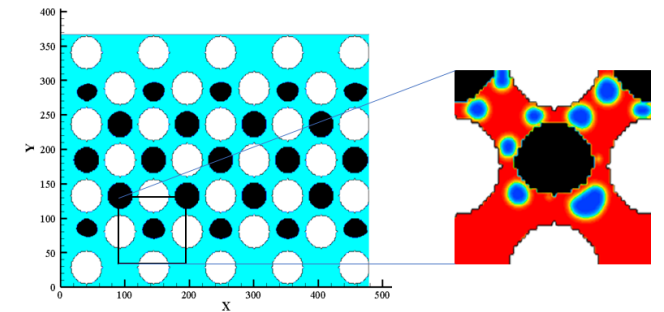


Grain-coating

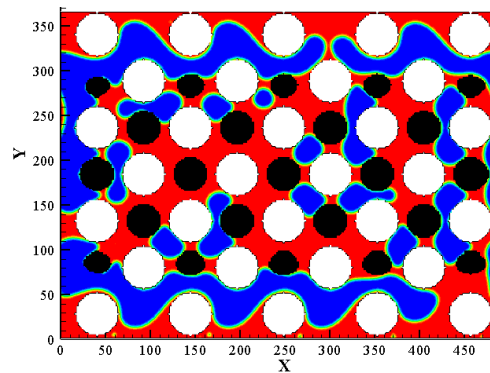
K_r vs S_w

Simulation for gas and water transport processes

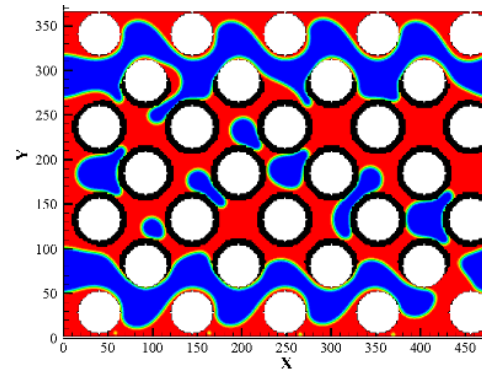
- Relative permeability estimation during hydrate dissociation
 - Thermal stimulation case: both considering latent heat and heat source
 - Top/bottom thermal intervention from hydrate-free layer was considered
 - Boundary conditions
 - Periodic boundary condition on outer boundaries
 - Bounce-back boundary condition on fluid-solid interfaces
 - Constant temperature boundary condition on top/bottom boundaries
 - Fully developed temperature boundary condition on inlet/outlet boundaries
 - Wettability boundary condition on solid boundaries



Gas generation during dissociation

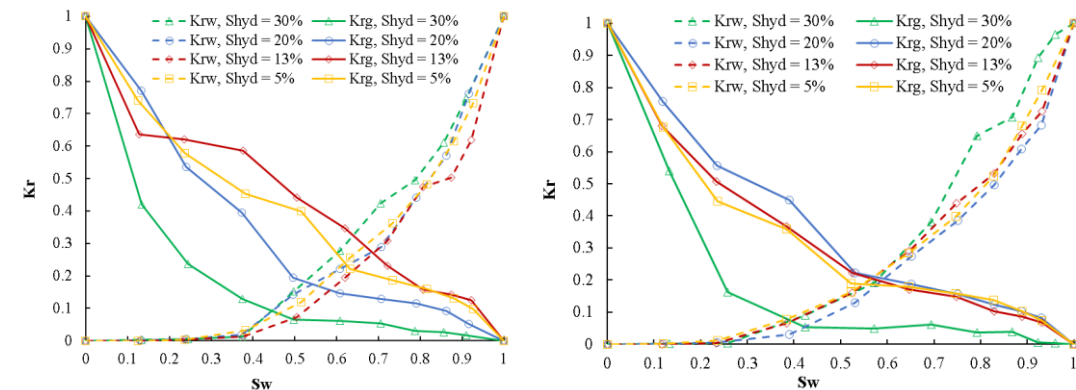


Pore filling



Grain-coating

$S_{hyd}=20\%$

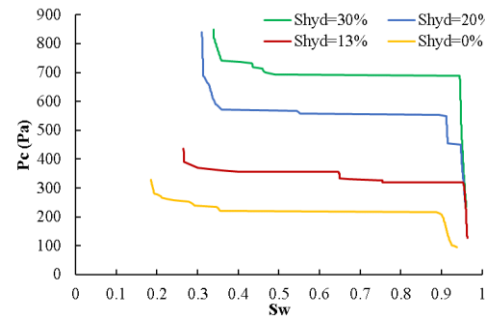


K_r vs S_w

Simulation for gas and water transport processes

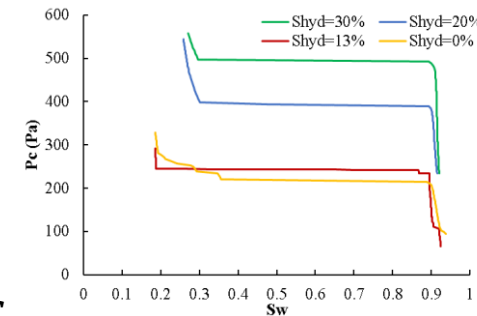
- P_C vs S_W relation comparisons for homogeneous cases

- P_C vs S_W relations under different hydrate saturation
- Boundary conditions
 - Constant pressure boundary condition on inlet boundary
 - Convective outflow boundary condition on outlet boundary



Pore-filling

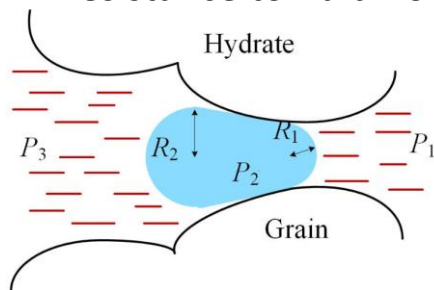
P_C vs S_W



Grain-coating

- Jamin effect

- resistance to fluid flow through capillaries which is due to the presence of bubbles/droplets

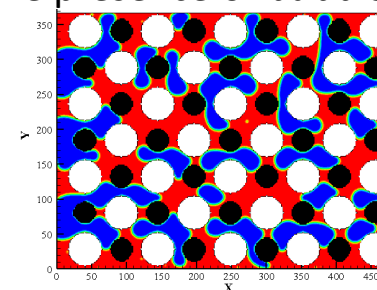


(Li et al., 2021)

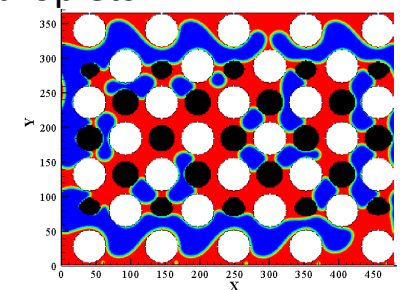
$$P_2 - P_1 = \frac{2\sigma_{wg}}{R_1}$$

$$P_2 - P_3 = \frac{2\sigma_{wg}}{R_2}$$

$$P_3 - P_1 = 2\sigma_{wg} \left(\frac{1}{R_1} - \frac{1}{R_2} \right)$$



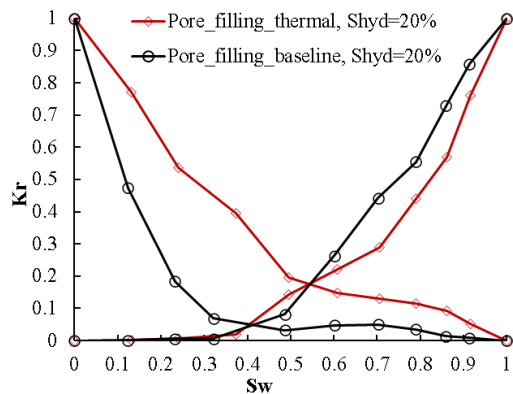
Baseline



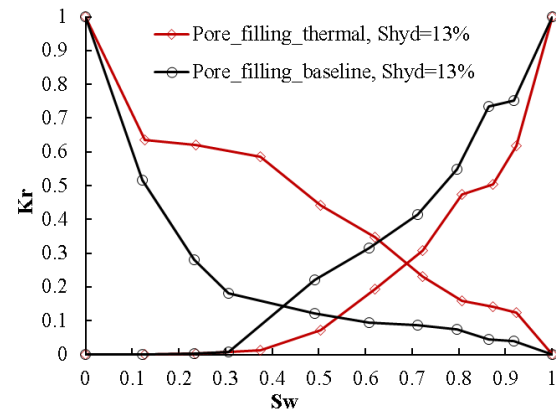
With thermal intervention

Simulation for gas and water transport processes

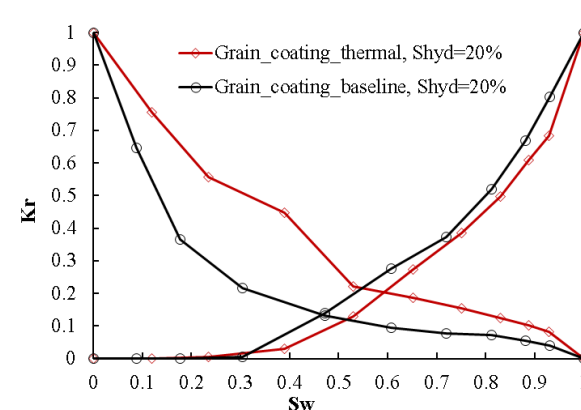
- K_r vs S_w relation comparisons for homogeneous cases
 - Gas gathers in areas without hydrates, while water remains in areas with hydrates due to capillarity
 - The non-wetting phase tends to occupy larger pores, whereas the wetting phase is more likely to occupy smaller pores
 - Gas exhibits lower dynamic viscosity than water, resulting in higher mobility
 - Capillary pressure and Jamin effect are significantly decreased
 - Thermal stimulation cases show considerable improvements in K_{rg} values but little decrease in K_{rw}



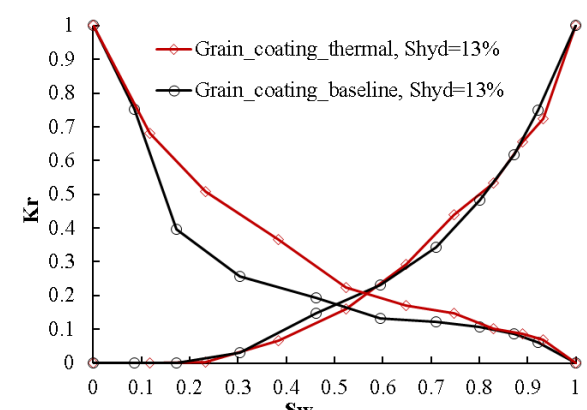
$S_{hyd}=20\%$



$S_{hyd}=13\%$



$S_{hyd}=20\%$



$S_{hyd}=13\%$

K_r vs S_w comparisons

Pore-filling

Grain-coating

Conclusion

- Concluding remarks
 - Uniform dissociation in baseline model obscures the formation of hydrate-free zone, making capillarity impact and Jamin effect on fluid transport dependent solely on S_{hyd}
 - The coupling of mass and heat transfer typically results in the creation of a hydrate-free zone, causing the redistribution of fluid under capillary effects, which significantly impacts the $K_r - S_{\text{hyd}}$ relation
 - Gas is prone to occupy the hydrate-free zone under capillary force as the non-wetting phase, leading to a significant increase in $K_{r,g}$ values compared to those obtained from baseline models
 - Considerable deviations can be found by comparing the $K_r - S_w$ relations obtained from coupled and baseline models
 - Gas bubble transport in HBS is mostly impeded by the Jamin effect, which prevents the formation of a continuous gas stream. Therefore, it is crucial to mitigate this effect during production to facilitate efficient gas extraction

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