



Simulation of reactive transport in heterogenous porous media with a Newton-Krylov method

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InterPore 2022 Abu-Dhabi, UAE (30 May - 2 June, 2022)

Motivations and scope of the talk

- Reactive transport = coupling of flow + transport with geochemistry,
- Applications: nuclear waste storage, CO2 or H2 underground storage.
- Variety of chemical reactions: aqueous (acid - base, redox), surface reactions (ion exchange), mineral precipitation dissolution

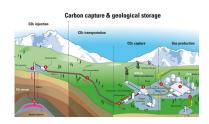


This talk

- One phase flow and transport, equilibrium chemistry
- Globally coupled approach that keeps chemistry and transport codes separate
- Solve mineral precipitation-dissolution with an interior point method
- Large non-linear system solved with a Newton–Krylov method. Nonlinear preconditioning ⇒ convergence independent of mesh size

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Chemical laws for equilibrium

Species: $(X_i^a)_{j=1,..,N_a}$ mobile, $(X_i^s)_{j=1,..,N_s}$ immobile, $(X_i^p)_{j=1,..,N_p}$ mineral

Chemical reactions

$$S\begin{pmatrix} X^{a} \\ X^{s} \\ X^{p} \end{pmatrix} \leftrightarrows \begin{pmatrix} 0 \\ 0 \\ 0 \end{pmatrix}, S = \begin{pmatrix} S_{aa} & 0 & 0 \\ S_{sa} & S_{ss} & 0 \\ S_{pa} & 0 & -I \end{pmatrix}$$

• N_r homogeneous reactions,
• \bar{N}_r heterogeneous reactions,
• N_p precipitation reactions.

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$$\begin{pmatrix} S_{aa} & 0 \\ S_{sa} & S_{ss} \end{pmatrix} \begin{pmatrix} \log c \\ \log s \end{pmatrix} = \begin{pmatrix} \log K_a \\ \log K_s \end{pmatrix}$$

$$\Pi = \log K_p - S_{pa} \log \mathbf{c}$$

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 c_i conc. species X_i^a , s_i conc. species X_i^s , p_i conc. species X_i^p

Aqueous and sorption reactions

Mass action law (log. form)

$$\begin{pmatrix} S_{aa} & 0 \\ S_{sa} & S_{ss} \end{pmatrix} \begin{pmatrix} \log \mathbf{c} \\ \log \mathbf{s} \end{pmatrix} = \begin{pmatrix} \log K_a \\ \log K_s \end{pmatrix}$$

Reactions involving minerals

Mass action law for minerals

 $(\Pi=0 \text{ and } p\geq 0) \text{ or } (\Pi\geq 0 \text{ and } p=0)$

Solubility product

$$\Pi = \log K_p - S_{pa} \log \mathbf{c}$$

Mass conservation with total concentrations

Mass conservation

$$\phi \partial_t \mathbf{c} + \mathcal{L} \mathbf{c} = S_{aa}^T r_a + S_{sa}^T r_s + S_{pa}^T r_p,$$

$$\rho_s (1 - \phi) \partial_t \mathbf{s} = S_{ss}^T r_s,$$

$$\rho_s (1 - \phi) \partial_t \mathbf{p} = -r_p.$$

 r_a, r_s, r_p : unknown reaction rates $\mathcal{L}(\mathbf{c}) = \nabla \cdot (q\mathbf{c} - D\nabla \mathbf{c})$ (advection-diffusion)

Elimination of the reaction rates ([Saaltink et al. WRR, 1998])

Introduce a kernel matrix U such that $US^T = 0$. Let $U = \begin{pmatrix} U_{aa} & U_{as} & U_{ap} \\ 0 & U_{ss} & 0 \end{pmatrix}$

We define : $C = U_{aa} c$, $\overline{C} = U_{as} s + U_{ap} p$, $T = C + \overline{C}$,

Transport for total concs.

$$\begin{split} \phi \partial_t & \overset{}{\mathsf{C}} + \rho_s (1 - \phi) \partial_t \overline{\mathsf{C}} + \mathcal{L} \overset{}{\mathsf{C}} &= 0, \\ \rho_s (1 - \phi) \partial_t \overline{\mathsf{T}} &= 0. \end{split}$$

Solution of the chemical sub-problem

Mass action laws

$$S_{aa} \log \mathbf{c} + \log K_a = 0$$

Definition of the totals

$$T = U_{aa}c + U_{pa}p$$

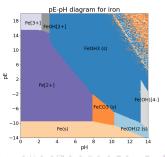
Complementarity condition

$$(\log \mathcal{K}_p - S_{pa}\xi^p = 0, p \geq 0) \text{ or } (\log \mathcal{K}_p - S_{pa}\xi^p \geq 0, p = 0).$$

- Nonlinear system with complementarity conditions
- Solve by Newton interior points method (F. Saaf (96)) or semi-smooth Newton (S. Kraütle (11))

Chemical solution operator

$$\Psi_{\mathsf{C}}: \mathsf{T} \mapsto \Psi_{\mathsf{C}}(\mathsf{T}) = \overline{\mathsf{C}} = U_{\mathsf{as}} \, \mathsf{s} + U_{\mathsf{ap}} \mathsf{p}.$$



Discrete advection-diffusion model



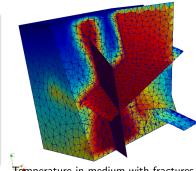
Discrete advection diffusion

$$A\mathbf{c}^{n+1} = B\mathbf{c}^n + M\Delta tq^{n+1}$$

Vertex Approximate Gradient scheme (Eymard et al., 2012)

ComPASS code (Lopez et al., 2018)

- Multiphase compositional thermal Darcy flow model (Coats formulation)
- 2D discrete fracture or fault network coupled with the surrounding 3D matrix
- 3D conforming polyhedral meshes
- Newton Raphson algorithm with phase appearance and disappearance
- CPR-AMG preconditioner [Lacroix et al, 2001], [Scheichl e al., 2003]



Temperature in medium with fractures

The coupled formulation

SIA: sequentially solve transport and chemistry, easy to code, slow convergence, loss of mass.

GIA: solve full coupled system, accurate, robust, difficult to code, needs efficient solver.

Chemistry is local \Rightarrow eliminate individual concentrations at each point by using the operator Ψ_C . Only unknowns : \mathbb{C} , $\bar{\mathbb{C}}$.

Coupled formulation

$$\phi \partial_t \mathbf{C} + \phi \partial_t \bar{\mathbf{C}} + \mathcal{L}(\mathbf{C}) = 0,$$

$$\bar{\mathbf{C}} - \Psi_C \left(\mathbf{C} + \bar{\mathbf{C}} \right) = 0.$$

Discrete non-linear system $(b^n = BC^n + M^{-n})$

$$f\begin{pmatrix} \mathbf{C} \\ \bar{\mathbf{C}} \end{pmatrix} = \begin{pmatrix} (A \otimes I)\mathbf{C} + (M \otimes I)\bar{\mathbf{C}} - \mathbf{b}^n \\ \bar{\mathbf{C}} - \Psi_{\mathbf{C}}(\mathbf{C} + \bar{\mathbf{C}}) \end{pmatrix} = 0$$

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Newton-Krylov with (non-linear) preconditioning

Inexact Newton method [Kelley, 1995])

- Solve the linear system by an iterative method (GMRES),
- GMRES requires only Jacobian matrix by vector products, can be computed analytically or approximated by finite differences.

Elimination of C

Reduced system
$$h(\bar{C}) = \bar{C} - \Psi_C \left((A^{-1} \otimes I) \left(b^n - (M \otimes I) \bar{C} \right) + \bar{C} \right) = 0.$$

Jacobian * vector $J_h v = v - J_C J_T v = v - J_C v + J_C \left((A^{-1}M) \otimes I \right) v.$

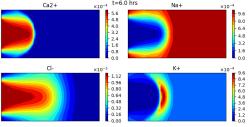
Details: [L. Amir, MK, IJNAM, 2019], [L. Amir, MK, Water, 2021]

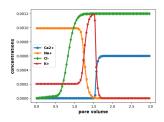
- Links between elimination and block preconditioning,
- The number of iterations becomes independent of the mesh.

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Ion exchange: 2D PhreeqC, Example 11

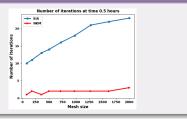
Inject calcium on part of left boundary into a K-Na solution



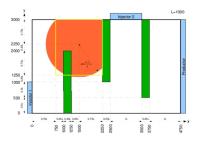


Fixed-point – precond. Newton-Krylov comparison

- Nb. of iterations \(\sqrt{} \) with the mesh resolution for SIA while it remains stable, small for NKM.
- NKM gives convergence independent of the mesh size.

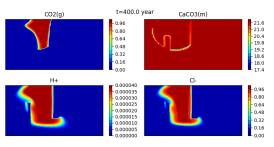


SHPCO2 Benchmark (precipitation) - IFPEN



Parameters (Heterogenous)

Barriers with low permeability Initial state: two zones with different initial concentrations, bubble contains CO2(g) equal to 1, 0 elsewhere.



Conclusion and perspectives

- Extension of the capabilities of the equilibrium chemical solver to handle mineral precipitation and dissolution;
- Heterogenous, 2D configurations: Ex11, SHPCO2
- Number of iterations is independent of the mesh size
- Improve the robustness of the NKM method in the presence of minerals.
- Improve the computation of the Jacobian matrix by vector product, by exploiting the known block structure of the matrix
- Extension to two-phase flow (cf [E. Ahusborde, M. Kern, V. Vostrikov, 2015], [E Ahusborde, B Amaziane, M El Ossmani, 2018])

Some references



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Preconditioning a coupled model for reactive transport in porous media

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Call for participation: two-phase reactive benchmark

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